Below Melting Point Photothermal Reshaping of Single Gold Nanorods Driven by Surface Diffusion

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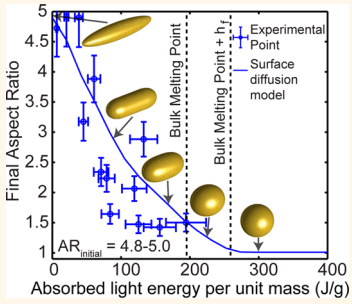
ABSTRACT Plasmonic gold nanorod instability and reshaping behavior below melting points are important for many future applications but are yet to be fully understood, with existing nanoparticle melting theories unable to explain the observations. Here, we have systematically studied the photothermal reshaping behavior of gold nanorods irradiated with femtosecond laser pulses to report that the instability is driven by curvature-induced surface diffusion rather than a threshold melting process, and that the stability dramatically decreases with increasing aspect ratio. We successfully utilized the surface diffusion model to explain the observations and found that the activation energy for surface diffusion was dependent on the aspect ratio of the rods, from 0.6 eV for aspect ratio of 5 to 1.5 eV for aspect ratio less than 3. This result indicates that the surface atoms are much easier to diffuse around in larger aspect ratio rods than in shorter rods and can induce reshaping at any given temperature. Current plasmonics and nanorod applications with the sharp geometric features used for greater field enhancement will therefore need to consider surface diffusion driven shape change even at low temperatures.



KEYWORDS: gold nanorods · photothermal reshaping · surface diffusion · plasmonics

The use of gold nanorods has become pervasive in many recent technological advancements including biolabeling,[1,2] sensing,[3,4] data storage,[5,6] photovoltaics,[7−9] and cancer therapy.[10,11] These techniques take advantage of the strong localized surface plasmon resonance (LSPR) inherent in the gold nanorods, and their stability is vital for preserving the LSPR and the success of these applications.[12−14] However, instability of the shape during laser exposure has been reported by many, and reshaping of nanorods was observed to occur at temperatures as low as 400 K.[15−17] Petrova *et al.* and Liu *et al.* observed thermal reshaping of aspect ratio 3 gold nanorods at low temperatures[15,17] when the nanorods were subjected to heat for an extended period. Khalavka *et al.*[18] reported similarly low-temperature shape transformation but also reported enhancing the stability by coating a thin carbon layer. Plech *et al.*[19] reported surface phase transition of gold nanoparticles at 400 K. Stability of gold nanorods during ultrafast pulsed laser irradiation is also of concern, due to the unclear reshaping threshold and the high temperature that can be reached during irradiation. Yamaguchi's group reported low aspect ratio gold nanoparticle reshaping at 5.6 mJ/cm^2 for 30 ps pulsed laser,[20] while Zijlstra *et al.*[5,21,22] reported 1.125 to 1.75 mJ/cm^2 for 100 fs pulsed laser irradiation.

Typically, the melting point of nanosized particles was predicted to be lower than the bulk melting point and was observed by many.[23−28] Melting models such as homogeneous melting and growth model,[26,27] liquid nucleation and growth,[23,29] liquid shell nucleation,[30,31] and liquid-drop (LD) models[28] have been proposed for spherical particles. For nanorods or wires, thermodynamic modeling[24] and LD models[32] were proposed to explain the reduced melting points with respect to aspect ratios of nanorods. However, these models were unsuccessful in explaining the observed instability and reshaping of nanorods at temperatures as low as 400 K.

Molecular dynamics simulations have been used to study melting behavior of gold nanorods with size much less than 500k atoms.[33−36] Wang and Dellago studied nanosecond pulse heating on gold nanorods of size less than 10k atoms to find lower melting points.[34,35] Recently Gan and

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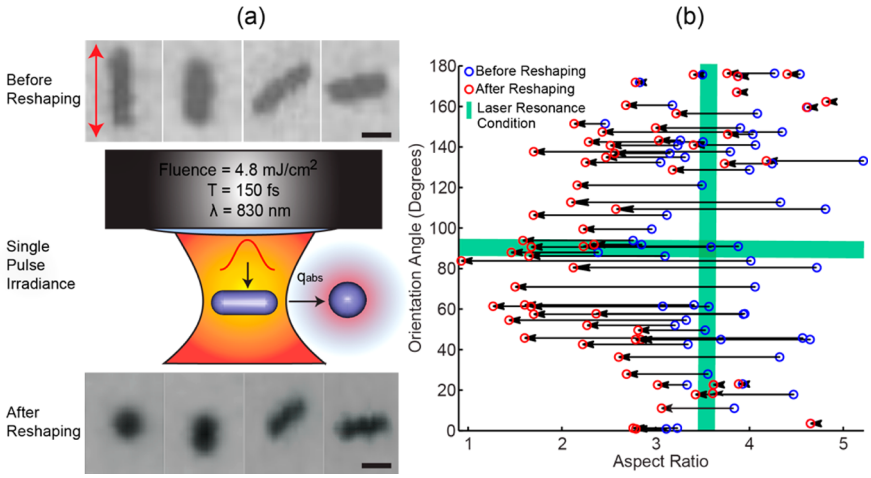

Figure 1. (a) Schematic showing the angular depletion method of photothermal reshaping: Nanorods are imaged on a coordinate-marked TEM slide in a precise location, then a single pulse is applied to all the nanorods to heat and reshape them, and finally, the same rods are located and imaged again to quantify the degree of reshaping. TEM image of the same nanorods before and after single pulse excitation with vertically polarized light. Scale bar is 20 nm. Note that as the orientation misalignment increases, the reshaping becomes less. (b) Vector plot showing reshaping trajectories of individual nanorods after laser pulse irradiation. Green lines show resonant rod aspect ratio/orientation with laser condition, *i.e.*, 830 nm wavelength and 90° polarization.

Jiang[36] extended the heating rate to a femtosecond time scale using a two-temperature model to match the recent experimental conditions of ultrafast laser exposures, which incorporates fast heat decay to the surrounding material. Surface melting was observed in these simulations. However, the limitation in simulation time scales of MD simulation prohibits probing complete evolution of rod shapes in an experimental time scale on the order of seconds. Typically, the MD simulation focuses on the instantaneous structures during rapid heating, which is far from the final structure that is being observed in microscopy.

Here, we report an experimental study on photothermal reshaping behavior of gold nanorods to show that their reshaping can be initiated well below the melting points and are heavily dependent on aspect ratio. From this result, we suggest curvature driven surface diffusion, rather than threshold melting, to be the main physical mechanism for photothermal reshaping of gold nanorods. We successfully explain the observations with a theoretical framework of surface diffusion by Mullins.[37−39] Previously, Herring[40] and Nichols and Mullins[37,41] used the curvature driven surface diffusion to explain the blunting of microscopic objects such as sharp scanning tunneling microscope tips, field emission guns, and sintering of metallic microspheres or smoothing of scratches on surfaces. More recently, Combe et al.[42] simulated the facet nucleation-induced shape change of nanoclusters (of less than 13 000 atoms) using a surface diffusion Monte Carlo method. The surface diffusion framework shown in these works allows low-temperature reshaping of gold nanorods without invoking the concept of low melting point. The time scales of previous reshaping experiments using constant temperature heating at low temperatures[15−17] were on the order of seconds and minutes, which is enough for surface diffusion to drive the shape change.

RESULTS AND DISCUSSION

Experimentally, we irradiated 600 randomly distributed and oriented gold nanorods (mean aspect ratio 3.5, average width 15.5 nm, from NanoSeedz) with linearly polarized, single femtosecond laser pulses (pulse width ∼150 fs) at 830 nm wavelength with a typical fluence of 4.8 mJ/cm^2 to "angularly deplete" the nanorods. The random distribution in nanorod aspect ratio and orientation angle, relative to laser wavelength and polarization, creates natural variation in absorbed energies between each rod, enabling the reshaping behavior to be extracted statistically, as a function of aspect ratio and absorbed energy. Further, excitation by ultrafast laser pulse induces a temperature sweep of the nanorods, allowing extraction of diffusion coefficients over a large temperature range rather than a single value. The activation energy for the diffusion could therefore be measured (full experimental details are provided in Supporting Information section 1 Materials and Methods and Figures S1−S7).

In Figure 1a, selected rods oriented at particular angles to laser pulse polarization are shown before and after the pulse irradiation to illustrate the variation in their absorbed energy and their reshaping. Nanorods with long axes aligned close to the polarization angle exhibit reshaping after laser irradiation, while the reshaping is markedly less for orthogonally aligned nanorods, as expected for $\cos^2\theta$ relationship of absorbed energy.[21] A scatter plot showing the orientation angle and aspect ratio of individual nanorods before and after reshaping is shown in Figure 1b, with the

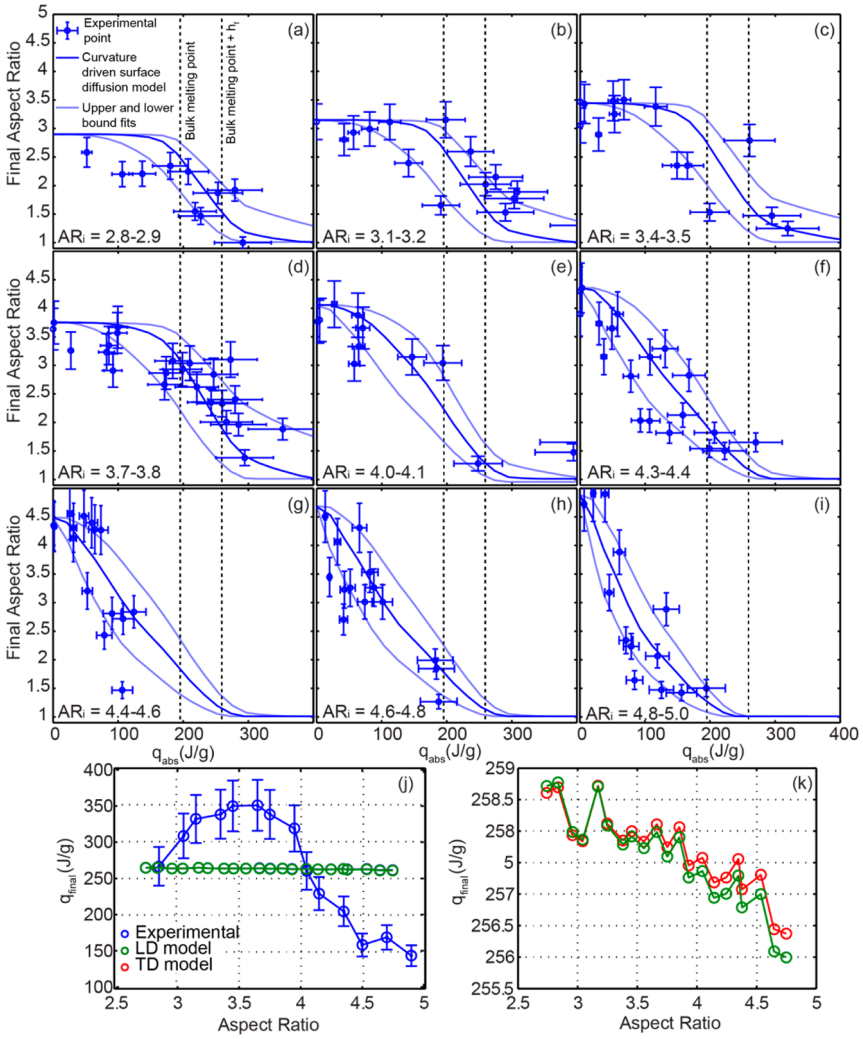

Figure 2. (a−i) Photothermal reshaping behavior of gold nanorods in respective aspect ratio range bins. Final aspect ratio *vs* absorbed energy density q_{abs}. Aspect ratio ranges from 2.7 to 5.0. Points with error bars are experimental results. Solid blue lines are the theoretical simulation using eqs 1−4; see text. The two vertical lines indicate the energy density needed to reach the melting temperature and to overcome the latent heat of fusion h_f. (j) Complete reshaping point q_{final} *vs* the aspect ratio extracted from the panels a−i. Black and red lines are the existing theoretical complete reshaping points, *i.e.*, q_{final}^{theory}. Thermodynamic model[24] and liquid-drop model[28,32] calculated for average aspect ratio and volume of the nanorods within each bin. (k) Close-up view of theory lines of q_{final}^{theory} values showing only a small change around 255 J/g.

arrows indicating the vectors for reshaping trajectories. One might expect nanorods with aspect ratios and orientation angles resonant to the laser pulse wavelength and polarization (crossing point of the green lines) to show the largest magnitude of the vector, and then it decreases as it moves away from the resonance position on the plot. However, such trend was not obvious from the figure. Instead, the nanorods with higher aspect ratios, that is, the right side of the vertical green line, show drastic reshaping toward spheres despite their reduced absorption cross sections at the laser wavelength, suggesting a decreasing thermodynamic stability with increasing aspect ratio.

This phenomenon is more clearly shown in Figure 2, where the final aspect ratios of the nanorods for a tight range of aspect ratio bins (0.1−0.2) are plotted against the absorbed energy from a single pulse. In order to account for distribution in the constituent nanorods' volume within each aspect ratio bin, the energy absorbed by each nanorod, Q_{abs}, is normalized by its mass m_{NR}, leading to the unit $q_{abs} = Q/m_{NR}$, the absorbed energy per unit mass in J/g. The energy density required to reach the bulk melting point of gold, q_{melt}^{bulk}, was estimated to be 195 J/g (calculated using the two-temperature model, TTM, with specific heat function of gold and heat loss, *i.e.*, $q_{melt}^{bulk} = \int_{T_i}^{T_f} C_{gold}(T)dT + q_{loss}$[43]) and that including the latent heat of fusion q_{final}^{bulk}, that is, $q_{final}^{bulk} = q_{melt}^{bulk} + (h_f + q_{loss})$ (~266 J/g), is shown in the figures as vertical lines. Examining Figure 2, we observe that nanorods undergo a gradual shape transition below the melting point instead of abrupt reshaping at the vertical lines, which might be expected for threshold melting. This indicates that the reshaping is not driven by threshold melting. In order to accurately compare the results to the existing theories of nanostructure melting point suppression, the

complete reshaping point, q_{final}^{expt} (*i.e.*, q for reaching final aspect ratio ~1), was extracted by fitting a simple quadratic function, ($AR_{final} = AR_{initial} - bq^2$) to the data points (fitting curves not shown).[44] Theoretical q_{final} values (*i.e.*, q_{final}^{theory}) from previous models for infinite wire melting points, thermodynamic (TD) model,[24] and LD model[28] with correction for finite nanorod aspect ratio according to the equation by Goswami *et al.*[32] are calculated for average sizes in the binning ranges (see details of those calculations in Supporting Information 2). The values are shown in Figure 2j,k, where the theory models only predict minimal variation for the rods used in this study. q_{final}^{expt} for an initial aspect ratio range from 2.7 to 4 (Figure 2a−e) is more or less similar to the previous result,[21] with the final reshaping to a sphere occurring about 20−70% above the energy density required to reach the bulk melting point and latent heat of fusion. However, as the aspect ratio increases (Figure 2f−i), the q_{final}^{expt} decreases dramatically, from 350 J/g for AR 3.6 to 150 J/g for AR 4.9. It is evident that for aspect ratios beyond 4, q_{final}^{expt} is well below that required to reach the melting point.

Such below melting point reshaping indicates that the existing theories for the melting point of nanorods are not sufficient to explain the observed results. Since the change in aspect ratio below the melting point has to involve mass diffusion and the nanorod surface has a large change in curvature from tip to the waist, the curvature driven surface diffusion is likely the driving mechanism for the observed reshaping.

Previously, Mullins[37−39] proposed a theory on the curvature driven reshaping and successfully explained the blunting of field emission gun tips in electron microscope or scanning tunneling microscope tips below the melting point. The central idea of this theory is the surface diffusion acting to minimize the surface energy of an object. The surface flux of diffusing atoms $\mathbf{J}_s$ can be expressed by

$$\mathbf{J}_s = \frac{\Omega v_s}{kT} \vec{\mathbf{D}}_s \cdot \nabla \mu \qquad (1)$$

where Ω is an atomic volume, v_s is the number of diffusing surface atoms in the unit area, k is the Boltzman's constant, T is temperature, and μ is the chemical potential. $\vec{\mathbf{D}}_s$ is the interface diffusivity tensor. The gradient notation ∇ is two-dimensional on the surface. Equation 1 should satisfy the equation of continuity, $(\partial n/\partial t) + \nabla \cdot \mathbf{J}_s = 0$, where $\partial n/\partial t$ represents the movement speed of a point on the surface in the outward normal direction to the surface. For an isotropic surface, the diffusivity tensor can be simplified as $D_s(T)$, with typical Arrhenius behavior with an activation energy E_a and a constant D_0 as

$$D_s(T) = D_0 \exp\left(\frac{-E_a}{kT}\right) \qquad (2)$$

The chemical potential μ can also be simplified for an isotropic surface, following the Herring's formula[37,40]

$$\mu = \mu_0 + \gamma_s \Omega K \qquad (3)$$

where $K = 1/R_x + 1/R_y$ is the mean curvature of the surface, γ_s is the free energy, and μ_0 is the chemical potential for a flat surface. Equation 1 becomes

$$\frac{dn}{dt} = v = B\nabla^2 K \qquad (4)$$

where $B = (\Omega^{4/3}\gamma_s D_s)/(kT)$. The equation relates the movement speed of a surface point to the curvature driven surface diffusion.

The finite difference method[37] can be applied to eq 4 to simulate the reshaping behavior. Movement of points on the entire surface of a nanorod is calculated using eq 4 for each time step $t = t_i$ to account for surface motion at given temperature T. Figure 3a and 3c show a reshaping evolution of an ellipsoidal nanorod (15 × 50 nm) subjected to a fixed temperature (430 K), which is an identical experimental condition that Liu *et al.* have previously shown.[17] We have therefore overlaid Liu *et al.*'s data on our curve in Figure 3c. One could see the gradual migration of atoms with respect to the time, from the high curvature (*i.e.*, tip) to the low curvature (*i.e.*, waist) of the rods. Good fit with the experimental results by Liu *et al.* shows that the theory accurately accounts the reshaping behavior at low temperatures. We have also independently fitted results by Petrova *et al.*[15] and Tollan *et al.*,[16] who used oven or heating stages to induce thermal reshaping of aspect ratio ~3.3 gold nanorods. The fitting parameter for these results revealed that the D_s varied between 10^{-10} and 10^{-15} cm^2/s depending on their temperature. This matches well with previously observed values for nanostructured gold surfaces.[45]

In photothermal reshaping using ultrafast pulsed laser irradiation, the temperature sweeps from room temperature to potentially above melting points in a picosecond time scale. In order to account for the temperature sweep in such a short time scale, we employed two-temperature model[43] (full details of TTM calculations are included in the Supporting Information 3) to calculate the temperature profile $T(t_i)$ of the gold nanorod during an ultrashort pulse irradiation. The fast temperature change with time steps on the order of 0.1 to 1 ps is accounted for in the $T(t_i)$, and it was input into eq 4 to simulate the reshaping evolution. The diffusion coefficient $D_s(T)$ is also a function of temperature, hence at each time step, the D_s value is obtained from the temperature $T(t_i)$ *via* the Arrhenius relation given by eq 2, with the activation energy, E_a, and the prefactor, D_0, used as fitting parameters.

A simulation of photothermal reshaping of a longer nanorod (AR 4.5), under ultrashort pulsed irradiation, is shown in Figure 3b for fixed values of E_a and prefactor

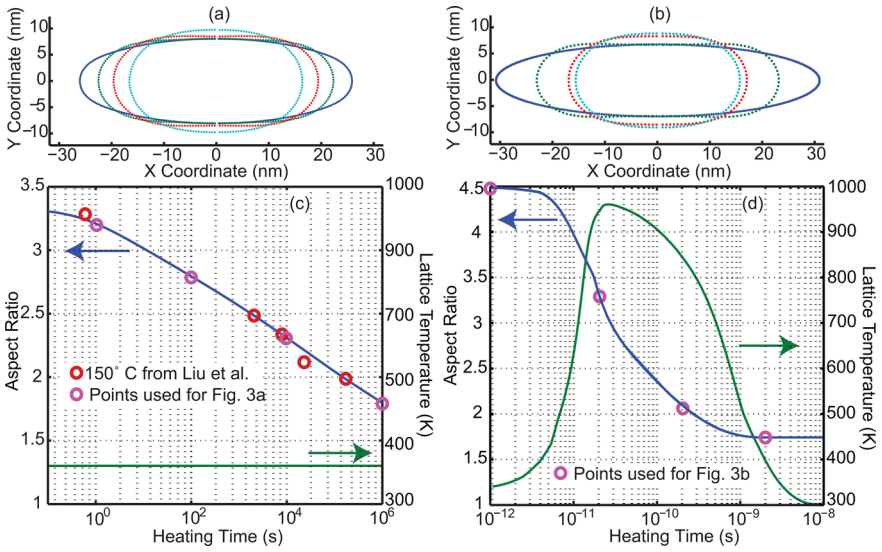

Figure 3. Simulated shape evolution of nanorods with aspect ratio 3.3 (a) for constant temperature at 430 K (~150 °C) for various times showing the reshaping trajectory. (c) Reshaping trajectory with respect to time (blue line) and the temperature is shown as a green line. Data by Liu *et al.*[17] are shown as red circles and show good agreement with the trajectory. The D_s = 2.12 × 10^{-12} cm^2/s was revealed from the fit. (b) Shape evolution of an aspect ratio 4.5 rod by a pulsed laser irradiation (150 fs pulse width). (d) Reshaping trajectory (blue line) and the temperature sweep profile, calculated using the TTM (see Supporting Information 3), and the reshaping profiles were calculated using eq 4. The activation energy E_a for the $D_s(T)$ curve was 0.75 eV.

D_0. The temperature profile is calculated using the TTM, with finite thermal conductance of the gold nanorod surface and surrounding poly(vinyl alcohol) accounted for. One could notice that, due to the fast temperature decay, the nanorod does not fully reshape into sphere, but rather "freezes" at a final aspect ratio of 1.7. This final value will obviously be dependent on absorbed energy density q_{abs}, and therefore, the observed reshaping into lower aspect ratio rods in Figure 2 can be accounted for using this theory.

This simulation technique was applied to fitting the observed reshaping of gold nanorods in Figure 2. By using eq 4 and E_a and D_0 as fitting factors, we evolved the nanorod shape in time to observe the final aspect ratio after no change was observed. Typically, the temperature falls back to room temperature within 10 ns, and reshaping slows to an undetectable rate. This provided a point in Figure 2, which plots the final reshaped aspect ratio AR_{final} against the absorbed energy, q_{abs}. This process was repeated by varying the pulse energy fluence F to obtain a complete curve for a single aspect ratio rod.

For comparison, we independently modeled the experimental conditions used by Liu *et al.*[17] in their experiment to extract the activation energy E_a and D_0 for AR 3.3 using eq 4. We chose Liu's results because their nanorods have similar sizes to the ones that we have used in the current study. We then used these values to fit experimental data for our rods. The D_0 value of 40 000 cm^2/s extracted from Liu *et al.*'s result was then used for other aspect ratio rods, assuming that at infinitely high temperature (*i.e.*, $1/T \sim 0$) the value should not be different for different aspect ratio. This allowed only E_a as a fitting factor with respect to the initial aspect ratio of nanorods, that is, $E_a(AR_{initial})$.

Other conditions for E_a during the fitting was that (1) if the temperature hits the bulk melting point, the diffusion coefficient would show a sudden jump to account for the latent heat of fusion, as observed by MD simulations;[46,47] (2) if the temperature overcomes the melting point (*i.e.*, $q_{abs} > q_{melt} + h_f$), the activation energy would be reduced to account for the liquid phase of matter; and (3) D_0 is allowed to vary during the reshaping to account for an increase in defects during reshaping. This was expressed according to the equation $D_0 = 40\,000 \exp(\Delta S_D/k)$, where ΔS_D is the change in the entropy in diffusion.[48] This is assumed to increase linearly on the change in aspect ratio due to amorphous configuration reached during reshaping.

Simulated theory lines with upper and lower fit bounds are shown in Figure 2 as solid lines, which show good agreement with the experimental points. The theory also accurately accounts for the reshaping behavior of nanorods with AR > 4, where the reshaping takes place at q_{final} values well below that required to reach the melting point. The reduced X^2 of the surface diffusion model is plotted for the aspect ratio bins, along with that of threshold model[44] based on TD/LD melting models for a comparison (Supporting Information 4, Figure S7). The X^2 is clearly superior for the surface diffusion model compared to threshold models.

The values of the parameters $D_s(T)$ lines and $E_a(AR_{initial})$ used for fitting individual AR values are shown in Figure 4. All the $D_s(T)$ lines (Figure 4a) show discontinuity at T = 1337 K to account for the latent heat of fusion. The $D_s(T)$ line for average AR 3.3 rods shows very good agreement with Liu *et al.*'s data

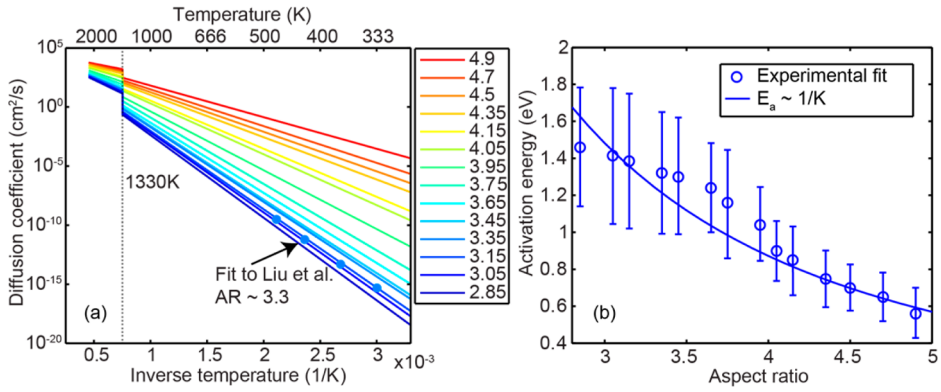

Figure 4. (a) Temperature-dependent surface diffusion coefficients $D_s(T)$ used for the theoretical fitting with eq 3 to the experimental data for a range of initial aspect ratios. The points indicate the diffusion coefficients used to fit Liu *et al.*'s experimental results.[17] (b) Activation energies used for the best fit to the experimental points in Figure 2. Overlaid is the $\sim 1/K$ activation energy relation with respect to aspect ratio, where K is the tip curvature.

(open circles). The activation energies needed to fit the experimental data (Figure 4b) show higher values for shorter aspect ratios. At AR 2.7, E_a is 1.45 eV, and then it decreases to 0.6 eV as the AR increases past 4.7. This means that, for higher aspect ratio rods, the energy barrier to allow surface atoms on the nanorods to be mobile and diffusive is significantly lowered, leading to easy reshaping well below the melting point. This also explains in Figure 2j why the shorter aspect ratio (AR < 4) rods have q_{final} well above those of the longer ones, therefore appearing to be "more difficult" to reshape than the longer ones.

Independent of the surface diffusion model, predicting the activation energy trend for nanorods requires boundary conditions where E_a at AR $\sim$ 1 (sphere) should be a finite value and E_a at AR $\sim \infty$ should converge to 0. A linear fit line fails to meet this condition, as it predicts a physically untenable negative value of E_a. Previously, the activation energy for the below melting point nucleation of crystal facets due to surface diffusion was predicted to follow $E_a \sim 1/K$.[42] In terms of the dimensions of the rods, the $E_a \sim 1/K$ curve has the following expression

$$E_a = \frac{2aC}{(AR^2 + 1)} \quad (5)$$

where a is the one-half nanorod length and C is a constant of proportionality (refer to Supporting Information 5 for the derivation). Applying a similar argument, the $E_a \sim 1/K$ curve for ellipsoidal nanorods is overlaid on Figure 4b. The trend is in a reasonable agreement with the experimental results, with an extracted value of C from Figure 4b of $C = 2.6 \times 10^8$ eV/m. It is noted that the curvature K is dependent on the aspect ratio and the dimensions of the rod (a in this case). Therefore, the aspect ratio dependencies of both curvature K and activation energy E_a have to be understood within the specified size regime.

Slightly larger E_a values (maximum ~20% at AR $\sim$ 3.5) than the trend are observed below AR < 4. This may be due to shape deviation from prolate spheroids to paraboloidally or hemispherically capped cylinders, which reduces the curvature around the tip and therefore increases E_a. This shape deviation should converge to the prolate spheroids for longer aspect ratio rods, as is observed in Figure 4b. Another potential source of error is the local temperature gradient within a single nanorod induced by a strong plasmon field, which would affect the chemical potential and therefore could disturb the purely curvature driven diffusion.

The absolute values of E_a are in good agreement with previous values in the literature. The reduced dimensions generally show reduced activation energy, at 1−1.5 eV for 40−50 nm thick gold films, that shows capillary-induced instability,[49] with a recent study showing an E_a less than 0.1 eV for ultrathin gold films (1−5 nm thick).[50] Activation energy for gold atomic steps is also shown to be in the range of 0.1−0.8 eV,[48] and for spherical gold nanoparticles with a diameter less than 5 nm, it is 0.54 eV.[51] With nanorod width ranging between 10 and 20 nm with many crystallographic facets, it is reasonable to observe the value between the published data.

Finally, we discuss the role of chemical potential in surface diffusion based reshaping and its interface with purely melting based reshaping. While the current surface diffusion model accounts for melting based reshaping by sudden increase in diffusion coefficient in the $D_s(T)$ at melting points, it does not differentiate melting based reshaping to purely surface diffusion driven reshaping. Furthermore, the current model does not account for surface diffusion by local surface melting, which may exist due to the different surface energies of crystal facets[33−35] or local temperature gradient, which may exist due to field enhancement at the tip during surface plasmon resonance.[52] To account for such complex situations, chemical potential must be rigorously treated.

The chemical potential expression used in this article ($\mu = \mu_0 + \gamma_s \Omega K$) has two terms.[37] The first term μ_0 is for

a flat surface and effectively contains all the temperature, pressure, and phase dependencies[53,54] (melting would be manifested in the phase differences). The second term is for a dependency with curvature only. The current formalism, therefore, does not take into account the surface diffusion due to spatial chemical potential gradients of temperature, pressure, or phase differences. Strong field enhancements at the tip of rods,[52] acoustic vibrations,[55] and patchy surface premelting[36] could all contribute to spatial gradients dependent on these parameters. If these gradients are assumed to exist within a nanorod, then the full expression for chemical potential is

$$\mu = \mu_T + \mu_{pr} + \mu_{ph} + \gamma_s \Omega K \quad (6)$$

and accordingly, eq 4 is now modified to incorporate the spatial gradients of these parameters

$$\frac{dn}{dt} = v = B_T \nabla^2 \mu_T + B_{pr} \nabla^2 \mu_{pr} + B_{ph} \nabla^2 \mu_{ph} + B_K \nabla^2 K \quad (7)$$

where B is the constant of proportionality and subscripts T, pr, and ph represent their dependence on temperature, pressure, and phase differences, respectively.

Determining individual components experimentally is beyond the scope of this paper. Perhaps variable temperature electron microscopy could be used to detect a sudden change in reshaping behavior near the melting point, and the terms μ_{ph} and B_{ph} might potentially be measured. In this case, a phase change (*i.e.*, melting) based reshaping process could be decoupled from surface diffusion based reshaping.

Proposed experimental work could also explain why the complete reshaping energy, q_{final}, at a certain aspect ratio of nanorods appears to match the existing theoretical melting models (*i.e.*, AR 2.8 in Figure 2j) better than others. We speculate that the amount of nanorod mass that has to be shifted to reshape into spheres plays an important role in explaining the apparent agreement with existing models at certain aspect ratios. For example, an aspect ratio 2.8 nanoparticle has 28% less mass to shift than the 3.5 nanorods to reshape to an aspect ratio 1 sphere. Here, two opposing processes exist, faster rates of diffusion for increasing aspect ratios and increasing amounts of mass which have to diffuse for these increased aspect ratios to reshape to spheres. There might be a crossing point for these two competing processes.

Study of these two competing effects will require detailed experimental study into shorter aspect ratios (1−3). Further, in order to remove the bias toward specific nanorod population selection, a transverse SPR could be excited (at ~530 nm), where all the rods have absorption.

CONCLUSION

In conclusion, we have shown that the thermal stability of gold nanorods dramatically decreases with increasing aspect ratio, and that reshaping can occur far below the bulk melting temperature for higher aspect ratios. We successfully explained the observation with surface diffusion on the nanorods, resulting from the increased surface curvature of the higher aspect ratio nanorods. The activation energy for the surface diffusion was found to be dependent on the aspect ratio, with values ranging from 0.6 to 1.5 eV. These findings will be especially important for the field of gold nanorod photothermal therapy and two-photon biolabeling, plasmonic circuitry, and solar cells using plasmonic structure,[1−4,7−11,14] where increased laser power and sharper geometric features are often seen as a mechanism for greater field enhancement. This work shows that surface diffusion based reshaping must be considered for plasmonic nanostructures for their stable operations, even at temperatures well below melting points.

METHODS

Gold nanorods were purchased from NanoSeedz (NR-20-750) which have an average aspect ratio of 3.5 (standard deviation 0.5) with an average width of 15.5 nm. From transmission electron microscope (TEM) images, we observe an aspect ratio ranging from 2.7 to 5, which is of interest to most applications. The nanorod solution was mixed with poly(vinyl alcohol) and was centrifuged on a TEM grid before the TEM imaging. Once the imaging was completed, the grid was transferred to an optical setup for laser irradiation.

The experimental setup for laser irradiation is shown in Figure S1. Single femtosecond laser pulses from a Ti:sapphire laser (Tsunami, Spectra-Physics) were picked using a pulse picker (Conoptics) and then focused on the sample using a 1.4 numerical aperture and an oil-immersion objective lens (Olympus). The sample was mounted onto a piezo stage for scanning. The 2D lattice of points used for heating is shown in Figure S2. The exposure was controlled using LabView, where a piezo stage moved the sample so that the shaded area was aligned with the full width at half-maximum focal spot of the 1.4 NA objective. This sample was again shifted, and the process was repeated until the entire sample area was exposed.

Laser scattering image is built up by confocal detection or backscattered photons from nanorods using a photomultiplier tube (PMT, Oriel), and the scattering cross sections, σ_{scat}^{meas}, for rods were extracted (Figure S3). Absorption cross sections of individual rods, σ_{abs}^{meas}, at 830 nm were estimated from the measured scattering cross sections, σ_{scat}^{meas}, using the equation $\sigma_{abs}^{meas}(\lambda = 830\ nm) = \sigma_{scat}^{meas}(\lambda = 830\ nm) \times \sigma_{abs}^{theory}/\sigma_{scat}^{theory}$. The σ_{abs}^{meas} values for ~600 nanorods are plotted in Figure S4 with theoretical absorption cross sections.

The variation in aspect ratio and orientation allowed variation in energy absorption per nanorods according to the equation $Q = \sigma_{abs} I \cos^2 \theta$, where θ is the angle between the polarization and rod orientation and subsequent photothermal reshaping. We imaged rods at single particle level using TEM before and

after the linearly polarized laser irradiation to observe their reshaping behavior. The geometry before and after reshaping is determined with the assistance of London Finder TEM grids, which use an alpha-numeric coordinate-marked system to locate the same area. To extract the geometric dimensions of the gold nanorods, we use the software package ImageJ, using the "Fit Ellipse" option to fit an ellipse around each particle and thus extract the length, width, and orientation angle. Typical wide area TEM micrographs are shown in Figure S5, with before and after laser irradiation images. A distribution in the length and widths of the 600 observed nanorods before reshaping is shown in Figure S6. A scatter plot showing the lengths and widths of the observed nanorods before and after melting is shown in Figure S7.

Conflict of Interest: The authors declare no competing financial interest.

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Supporting Information Available: Detailed experimental methods (Figures S1−S8), nanorod melting theories, and a two-temperature model is provided. This material is available free of charge *via* the Internet at http://pubs.acs.org.

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